

NEET 2024 QUESTION PAPER (CODE-Q6)

Test Booklet code

ENGLISH
Q6

This Booklet contains 32 pages, including Rough Page.

Do not open this Test Booklet until you are asked to do so.

NEET-UG 2025



Important Instructions:

1. The test is of **3 hours 20 minutes** duration and the Test Booklet contains **200** multiple-choice questions (four options with a single correct answer) from **Physics, Chemistry and Biology (Botany and Zoology)**. **50 questions** in each subject are divided into **two Sections (A and B)** as per details given below:
 - (a) **Section A** shall consist of **35 (Thirty-five)** Questions in each subject (Question Nos - 1 to 35, 51 to 85, 101 to 135 and 151 to 185). All questions are compulsory.
 - (b) **Section B** shall consist of **15 (Fifteen)** questions in each subject (Question Nos - 36 to 50, 86 to 100, 136 to 150 and 186 to 200). In Section B, a candidate needs to **attempt any 10 (Ten)** questions out of **15 (Fifteen)** in each subject.**Candidates are advised to read all 15 questions in each subject of Section B** before they start attempting the question paper. In the event of a candidate attempting more than ten questions, **the first ten questions answered by the candidate shall be evaluated.**
2. Each question carries **4** marks. For each correct response, the candidate will get **4** marks. For each incorrect response, one mark will be deducted from the total scores. **The maximum marks are 720.**
3. Use **Blue/Black Ball Point Pen only** for writing particulars on this page/ marking responses on Answer Sheet.
4. **The CODE for this Booklet is T2. Make sure that the CODE printed on the Original Copy of the Answer Sheet is the same as that on this Test Booklet.** In case of discrepancy, the candidate should immediately report the matter to the Invigilator for replacement of both the Test Booklet and the Answer Sheet.
5. The candidates should ensure that the Answer Sheet is not folded. Do not make any stray marks on the Answer Sheet. Do not write your Roll No. anywhere else except in the specified space in the Test Booklet/ Answer Sheet.
6. Use of white fluid for correction is **NOT** permissible on the Answer Sheet.
7. Each candidate must show on-demand his/her Admit Card to the Invigilator.
8. No candidate, without special permission of the centre Superintendent or Invigilator, would leave his/her seat.
9. The candidates should not leave the Examination Hall without handing over their Answer Sheet to the Invigilator on duty and sign (with time) the Attendance Sheet **twice. Cases, where a candidate has not signed the Attendance Sheet second time, will be deemed not to have handed over the Answer Sheet and dealt with as an Unfair Means case.**
10. Use of Electronic/Manual Calculator is prohibited.
11. The candidates are governed by all Rules and Regulations of the examination with regard to their conduct in the Examination Room/Hall. All cases of unfair means will be dealt with as per the Rules and Regulations of this examination.
12. **No part of the Test Booklet and Answer Sheet shall be detached under any circumstances.**
13. The candidates will write the Correct Test Booklet Code as given in the Test Booklet/ Answer Sheet in the Attendance Sheet.

Compensatory time of one hour five minutes will be provided for the examination of three hours and 20 minutes duration, whether such candidate (having a physical limitation to write) uses the facility of Scribe or not.

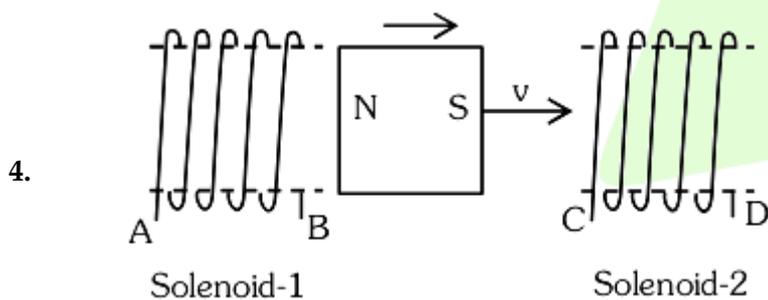
Section -A

1. A bob is whirled in a horizontal plane by means of a string with an initial speed of ω rpm. The tension in the string is T . If speed becomes 2ω while keeping the same radius, the tension in the string becomes:
- (A) T (B) $4T$ (C) $\frac{T}{4}$ (D) $\sqrt{2}T$
2. A particle moving with uniform speed in a circular path maintains:
- (A) constant velocity
 (B) constant acceleration.
 (C) constant velocity but varying acceleration
 (D) varying velocity and varying acceleration
3. A logic circuit provides the output Y as per the following truth table:

A	B	Y
0	0	1
0	1	0
1	0	1
1	1	0

The expression for the output Y is

- (A) $A \cdot B + \bar{A}$ (B) $A \cdot \bar{B} + \bar{A}$ (C) \bar{B} (D) B



In the above diagrams, a strong bar magnet is moving towards solenoid-2 from solenoid-1. The direction of induced current in solenoid-1 and that in solenoid-2, respectively, are through the directions:

- (A) AB and DC (B) BA and CD
 (C) AB and CD (D) BA and DC

5. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**.

Assertion (A) :- The potential (V) at any axial point, at 2 m distance (r) from the centre of the dipole of dipole moment vector \vec{P} of magnitude, $4 \times 10^{-6} \text{Cm}$, is $\pm 9 \times 10^3 \text{ V}$

(Take $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ SI Units}$)

Reason (R) :- $V = \pm \frac{2P}{4\pi\epsilon_0 r^2}$, where r is the distance of any axial point, situated at 2 m from the centre of the dipole.

In the light of the above statements. choose the correct answer from the options given below:

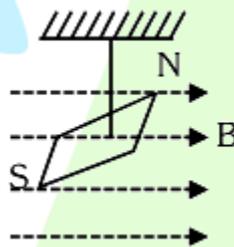
- (A) Both A and R are true and R is the correct explanation of A.
 (B) Both A and R are true and R is NOT the correct explanation of A.
 (C) A is true but R is false.
 (D) A is false but B is true.

6. Match **List-I** with **List-II**

	List-I		List-II
A.	Diamagnetic	I.	$\chi = 0$
B.	Ferromagnetic	II.	$0 > \chi \geq -1$
C.	Paramagnetic	III.	$\chi \gg 1$
D.	Non-Magnetic &	IV.	$0 < \chi < \epsilon$ (a small positive number)

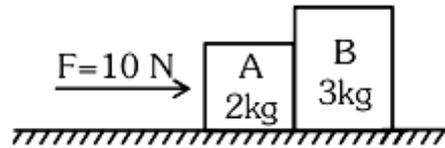
Choose the correct answer from the options given below:

- (A) A-II, B-III, C-IV, D-I
 (B) A-II, B-I, C-III, D-IV
 (C) A-III, B-II, C-I, D-IV
 (D) A-IV, B-III, C-II, D-I
7. In a uniform magnetic field of 0.049 T. a magnetic needle performs 20 complete oscillations in 5 seconds as shown. The moment of inertia of the needle is $9.8 \times 10^{-6} \text{kg m}^2$. If the magnitude of magnetic moment of the needle is $x \times 10^{-5} \text{ Am}^2$; then the value of 'x' is:

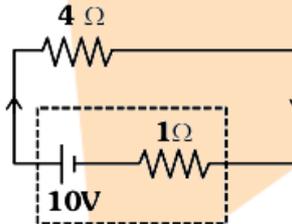


- (A) $5\pi^2$ (B) $128\pi^2$ (C) $50\pi^2$ (D) $1280\pi^2$
8. In an ideal transformer, the turns ratio $\frac{N_p}{N_s} = \frac{1}{2}$. The ratio $V_s : V_p$ is equal to (the symbols carry their usual meaning) :
- (A) 1: 2 (B) 2: 1 (C) 1: 1 (D) 1: 4
9. In a vernier calipers, (N+1) divisions of vernier scale coincide with N divisions of main scale. If 1 MSD represents 0.1 mm, the vernier constant (in cm) is:
- (A) $\frac{1}{10N}$ (B) $\frac{1}{100(N+1)}$ (C) 100 N (D) 10(N+1)

10. A horizontal force 10 N is applied to a block A as shown in figure. The mass of blocks A and B are 2 kg and 3 kg, respectively. The blocks slide over a frictionless surface. The force exerted by block A on block B is :



- (A) zero (B) 4N (C) 6N (D) 10N
11. If $x = 5 \sin\left(\pi t + \frac{\pi}{3}\right)$ m particle executing simple harmonic motion. The amplitude and time period of motion respectively, are :
- (A) 5cm, 2s (B) 5m, 2s (C) 5cm, 1s (D) 5m, 1 s
12. The terminal voltage of the battery, whose emf is 10V and internal resistance 1Ω , when connected through an external resistance of 4Ω as shown in the figure



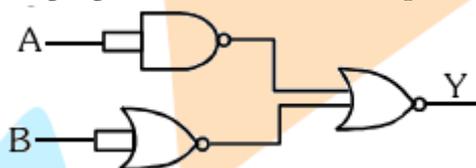
- (A) 4V (B) 6V (C) 8V (D) 10V
13. Given below are two statements :
- Statement I :** Atoms are electrically neutral as they contain equal number of positive and negative charges.
- Statement II :** Atoms of each element are stable and emit their characteristic spectrum. In the light of the above statements, choose the most appropriate answer from the options given below :
- (A) Both Statement I and Statement II are correct.
 (B) Both Statement I and Statement II are incorrect.
 (C) Statement I is correct but Statement II is incorrect.
 (D) Statement I is incorrect but Statement II is correct
14. If c is the velocity of light in free space, the correct statements about photon among the following are :
- A. The energy of a photon is $E = h\nu$
 B. The velocity of a photon is c .
 C. The momentum of a photon, $p = \frac{h\nu}{c}$
 D. In a photon-electron collision, both total energy and total momentum are conserved. E. Photon possesses positive charge.
- Choose the correct answer from the options given below :
- (A) A and B only (B) A, B, C and D only
 (C) A, C and D only (D) A, B, D and E only

15. Match List I with List II.

	List I		List-II
	(Spectral Lines of Hydrogen for transition for transitions from		(Wavelength (nm))
A.	$n_2 = 3$ to $n_1 = 2$	I.	410.2
B.	$n_2 = 4$ to $n_1 = 2$	II.	434.1
C.	$n_2 = 5$ to $n_1 = 2$	III.	656.3
D.	$n_2 = 6$ to $n_1 = 2$	IV.	486.1

Choose the correct answer from the options given below:

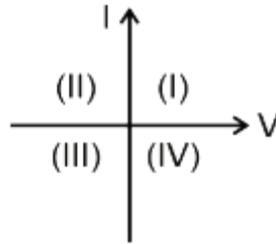
- (A) A-II, B-I, C-IV, D-III
 (B) A-III, B-IV, C-II, D-I
 (C) A-IV, B-III, C-I, D-II
 (D) A-I, B-II, C-III, D-IV
16. A tightly wound 100 turns coil of radius 10 cm carries a current of 7A. The magnitude of the magnetic field at the centre of the coil is (Take permeability of free space as $4\pi \times 10^{-7}$ (units):
 (A) 44 mT (B) 4.4 T (C) 4.4 mT (D) 44 T
17. The output (Y) of the given logic gate is similar to the output of an/a :



- (A) NAND gate (B) NOR gate (C) OR gate (D) AND gate
18. A wire of length 'l' and resistance 100Ω is divided into 10 equal parts. The first 5 parts are connected in series while the next 5 parts are connected in parallel. The two combinations are again connected in series. The resistance of this final combination is:
 (A) 26Ω (B) 52Ω (C) 55Ω (D) 60Ω
19. ${}_{82}^{290}\text{X} \xrightarrow{\alpha} \text{Y} \xrightarrow{e^+} \text{Z} \xrightarrow{\beta^-} \text{P} \xrightarrow{e^-} \text{Q}$
 In the nuclear emission stated above, the mass number and atomic number of the product Q respectively, are:
 (A) 280,81 (B) 286,80 (C) 288,82 (D) 286.81
20. The maximum elongation of a steel wire of 1 m length if the elastic limit of steel and its Young's modulus, respectively, are $8 \times 10^8 \text{ N m}^{-2}$ and $2 \times 10^{11} \text{ Nm}^{-2}$ is :
 (A) 4 mm (B) 0.4 mm (C) 40 mm (D) 8 mm
21. If the monochromatic source in Young's double slit experiment is replaced by white light, then
 (A) interference pattern will disappear.
 (B) there will be a central dark fringe surrounded by a few coloured fringes.
 (C) there will be a central bright white fringe surrounded by a few coloured fringes.
 (D) all bright fringes will be of equal width.

22. At any instant of time t , the displacement of any particle is given by $2t - 1$ (SI unit) under the influence of force of 5 N . The value of instantaneous power is (in SI unit):
 (A) 10 (B) 5 (C) 7 (D) 6

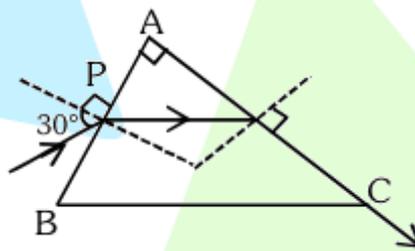
23. Consider the following statements A and B and identify the correct answer:



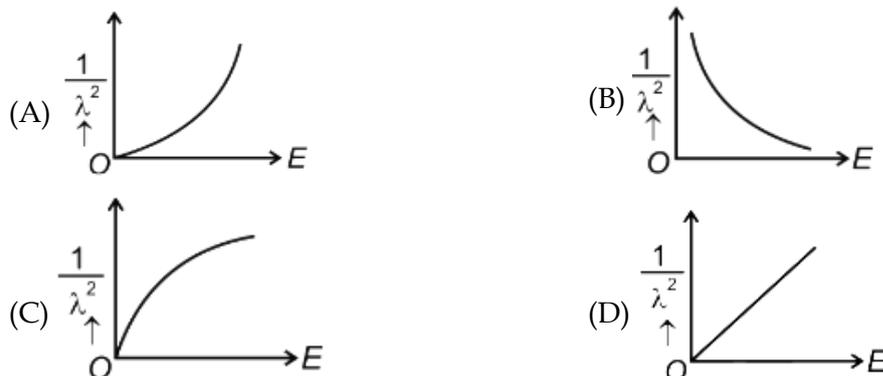
A. For a solar-cell, the I-V characteristics lies in the IV quadrant of the given graph.

B. In a reverse biased pn junction diode, the current measured in (μA), is due to majority charge carriers.

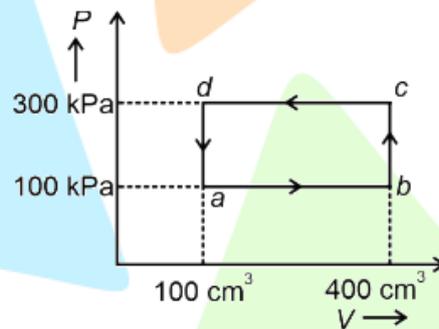
- (A) A is correct but B is incorrect
 (B) A is incorrect but B is correct
 (C) Both A and B are correct
 (D) Both A and B are incorrect
24. Two bodies A and B of same mass undergo completely inelastic one dimensional collision. The body A moves with velocity v_1 while body B is at rest before collision. The velocity of the system after collision is v_2 . The ratio $v_1 : v_2$ is:
 (A) 1: 2 (B) 2: 1 (C) 4: 1 (D) 1: 4
25. A light ray enters through a right angled prism at point P with the angle of incidence 30° as shown in figure. It travels through the prism parallel to its base BC and emerges along the face AC. The refractive index of the prism is:



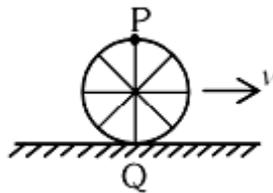
- (A) $\frac{\sqrt{5}}{4}$ (B) $\frac{\sqrt{5}}{2}$ (C) $\frac{\sqrt{3}}{4}$ (D) $\frac{\sqrt{3}}{2}$
26. The graph which shows the variation of $\left(\frac{1}{\lambda^2}\right)$ and its kinetic energy. E is



27. The quantities which have the same dimensions as those of solid angle are:
 (A) strain and angle (B) stress and angle
 (C) strain and arc (D) angular speed and stress
28. An unpolarised light beam strikes a glass surface at Brewster's angle. Then :-
 (A) the reflected light will be partially polarised.
 (B) the refracted light will be completely polarised.
 (C) both the reflected and refracted light will be completely polarised.
 (D) the reflected light will be completely polarised but the refracted light will be partially polarised.
29. The moment of inertia of a thin rod about an axis passing through its mid point and perpendicular to the rod is 2400 g cm^2 . The length of the 400 g rod is nearly:
 (A) 8.5 cm (B) 17.5 cm (C) 20.7 cm (D) 72.0 cm
30. A thin flat circular disc of radius 4.5 cm is placed gently over the surface of water. If surface tension of water is 0.07 Nm^{-1} , then the excess force required to take it away from the surface is :
 (A) 19.8 mN (B) 198 N (C) 1.98 mN (D) 99 N
31. A thermodynamic system is taken through the cycle aboda. The work done by the gas along the path bc is :



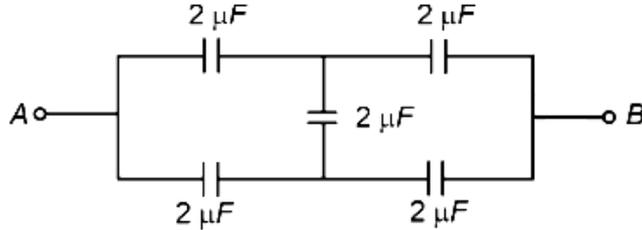
- (A) Zero (B) 30 J (C) -90 J (D) -60 J
32. A wheel of a bullock cart is rolling on a level road as shown in the figure below. If its linear speed is v in the direction shown, which one of the following options is correct (P and Q are any highest and lowest points on the wheel, respectively)?



- (A) Point P moves slower than point Q.
 (B) Point P moves faster than point Q.
 (C) Both the points P and Q move with equal speed.
 (D) Point P has zero speed.

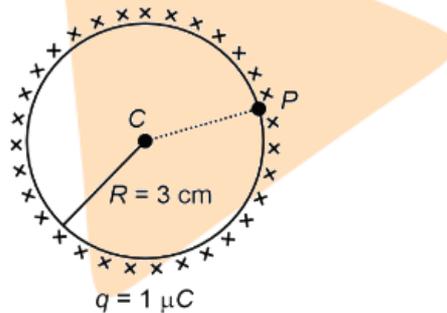
33. The mass of a planet is $\frac{1}{10}$ th that of the earth and its diameter is half that of the earth. The acceleration due to gravity on that planet is:
 (A) 19.6 ms^{-2} (B) 9.8 ms^{-2} (C) 4.9 ms^{-2} (D) 3.92 ms^{-2}

34. In the following circuit, the equivalent capacitance between terminal A and terminal B is:



- (A) $2 \mu\text{F}$ (B) $1 \mu\text{F}$ (C) $0.5 \mu\text{F}$ (D) $4 \mu\text{F}$

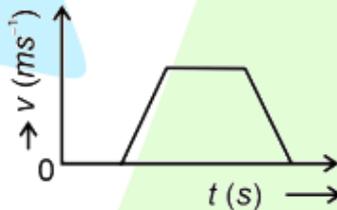
35. A thin spherical shell is charged by some source. The potential difference between the two points C and P (in V) shown in the figure is :
 (Take $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ SI units}$)



- (A) 3×10^5 (B) 1×10^5 (C) 0.5×10^5 (D) Zero

Section -B

36. The velocity (v) - time (t) plot of the motion of a body is shown below :



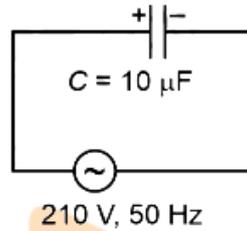
The acceleration (a) - time (t) graph that best suits this motion is :

- (A) (B) (C) (D)

37. If the mass of the bob in a simple pendulum is increased to thrice its original mass and its length is made half its original length, then the new time period of oscillation is $\frac{x}{2}$ times its original time period. Then the value of x is:

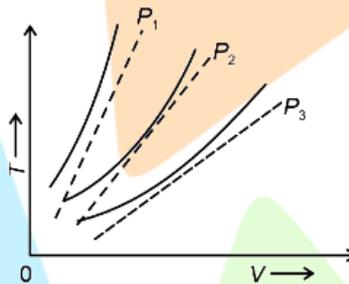
(A) $\sqrt{3}$ (B) $\sqrt{2}$ (C) $2\sqrt{3}$ (D) 4

38. A $10\mu\text{F}$ capacitor is connected to a 210 V, 50 Hz source as shown in figure. The peak current in the circuit is nearly ($\pi = 3.14$):



(A) 0.58A (B) 0.93A (C) 1.20A (D) 0.35

39. The following graph represents the T-V curves of an ideal gas (where T is the temperature and V is volume) at three pressures P_1 , P_2 and P_3 compared with those of Charles's law represented as dotted lines.



Then the correct relation is:

(A) $P_3 > P_2 > P_1$ (B) $P_1 > P_3 > P_2$
 (C) $P_2 > P_1 > P_3$ (D) $P_1 > P_2 > P_3$

40. An iron bar of length L has magnetic moment M . It is bent at the middle of its length such that the two arms make an angle 60° with each other. The magnetic moment of this new magnet is:
41. The minimum energy required to launch a satellite of mass m from the surface of earth of mass M and radius R in a circular orbit at an altitude of $2R$ from the surface of the earth is:

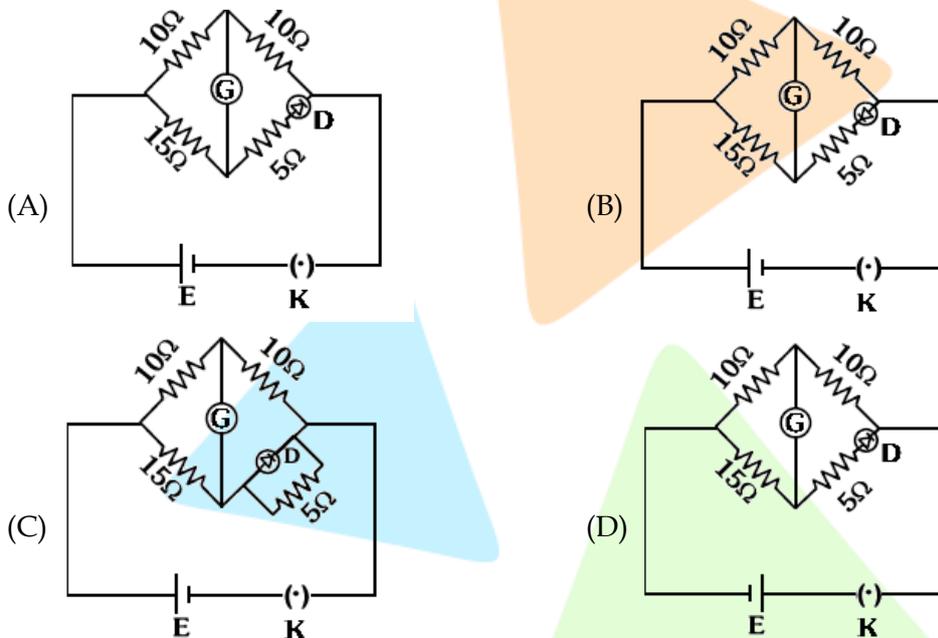
(A) $\frac{5CmM}{6R}$ (B) $\frac{2GmM}{3R}$ (C) $\frac{GmM}{2R}$ (D) $\frac{GmM}{3R}$

42. A parallel plate capacitor is charged by connecting it to a battery through a resistor. If I is the current in the circuit, then in the gap between the plates:

(A) there is no current.
 (B) displacement current of magnitude equal to I flows in the same direction as I .
 (C) displacement current of magnitude equal to I flows in a direction opposite to that of I .
 (D) displacement current of magnitude greater than I flows but can be in any direction.

43. The property which is not of an electromagnetic wave travelling in free space is that :
- (A) they are transverse in nature.
 (B) the energy density in electric field is equal to energy density in magnetic field.
 (C) they travel with a speed equal to $\frac{1}{\sqrt{\mu_0\epsilon_0}}$
 (D) they originate from charges moving with uniform speed.
44. A metallic bar of Young's modulus, $0.5 \times 10^{11} \text{ Nm}^{-2}$ and coefficient of linear thermal expansion 10^5 C^{-1} , length 1 m and area of cross-section 10^{-3} m^2 is heated from 0°C to 100°C without expansion or bending. The compressive force developed in it is:
- (1) $5 \times 10^3 \text{ N}$ (2) $50 \times 10^3 \text{ N}$ (3) $100 \times 10^3 \text{ N}$ (4) $2 \times 10^3 \text{ N}$

45. Choose the correct circuit which can achieve the bridge balance



46. A sheet is placed on a horizontal surface in front of a strong magnetic pole. A force is needed to:
- A. hold the steel there if it is magnetic.
 B. hold the steel there if it is non-magnetic.
 C. move the steel away from the pole with uniform velocity if it is conducting.
 D. move the steel away from the pole with uniform velocity if it is both, non-conducting and non-polar.
- Choose the correct statement(s) from the options given below:
- (A) B and D only
 (B) A and C only
 (C) A, C and D only
 (D) C only

53. Given below are two statements :

Statement-I : The boiling point of hydrides of Group-16 elements follow the order $H_2O > H_2Te > H_2Se > H_2S$.

Statement-II : On the basis of molecular mass.

H_2O is expected to have lower boiling point than the other members of the group but due to the presence of ex bonding in H_2O , it has higher boiling point.

In the light of the above statements, choose the correct answer from the options given below:

- (A) Both statement-I and Statement-II are true.
 (B) Both statement-I and Statement-II are false.
 (C) Statement-I is the true but Statement-II is false.
 (D) Statement-I is false but Statement-II is true.

54. Match List I with List II.

List I (Compound)	List II (Shape/geometry)
(A) NH_3	(I) Trigonal Pyramidal
(B) BrF_5	(II) Square Planar
(C) XeF_4	(III) Octahedral
(D) SF_6	(IV) Square Pyramidal

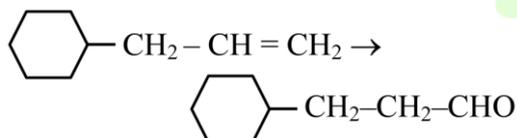
Choose the correct answer from the options given below:

- (A) A-I, B-IV, C-II, D-III
 (B) A-II, B-IV, C-III, D-I
 (C) A-III, B-IV, C-I, D-II
 (D) A-II, B-III, C-IV, D-I

55. The highest number of helium atoms is in:

- (A) 4 mol of helium
 (B) 4 u of helium
 (C) 4 g of helium
 (D) 2.271098 L of helium at STP

56. Identify the correct reagents that would bring about the following transformation

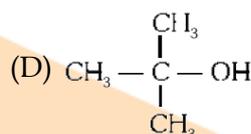
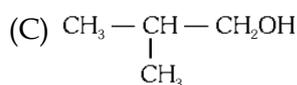
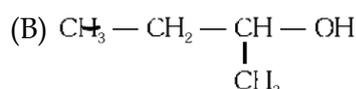
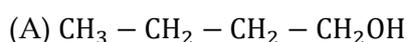


- (A) (i) $\frac{H_2O}{I^+}$ (ii) CrO_3
 (B) (i) BH_3 (ii) H_2O_2/OH^\ominus (iii) PCC
 (C) (i) BH_3 (ii) $\frac{H_2O_2}{OH^\ominus}$ (iii) Alk. $KMnO_4$ (iv) H_3O^\oplus
 (D) (i) H_2O/H^+ (ii) PCC

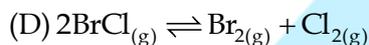
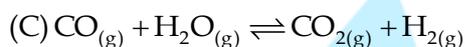
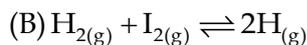
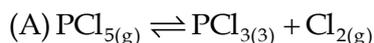
57. Match List I with List II.

List-I (Process)		List-II (Conditions)	
A.	Isothermal process	I.	No heat exchange
B.	Isochoric process	II.	Carried out at constant temperature
C.	Isobaric process	III.	Carried out at constant volume
D.	Adiabatic process	IV.	Carried out at constant pressure

58. Which one of the following alcohols reacts instantaneously with Lucas reagent?



59. In which of the following equilibria, K_p and K_c are NOT equal?



60. Match List I with List II.

List I

Quantum Number

A. m_l

B. m_s

C. l

D. n

List II

Information provided

I. shape of orbital

II. size of orbital

III. orientation of orbital

IV. orientation of spin of electron

Choose the correct answer from the options given below:

(A) A-I, B-III, C-II, D-IV

(B) A-III, B-IV, C-I, D-II

(C) A-III, B-IV, C-II, D-I

(D) A-II, B-I, C-IV, D-III

61. Given below are two statements:

Statement I : Aniline does not undergo Friedel Crafts alkylation reaction

Statement II : Aniline cannot be prepared through Gabriel synthesis.

In the light of the above statements, choose the correct answer from the options given below:

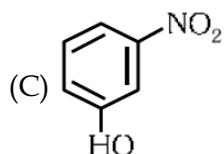
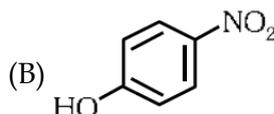
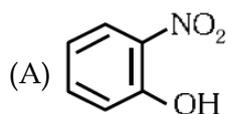
(A) Both Statement I and Statement II are true.

(B) Both Statement I and Statement II are false.

(C) Statement I is correct but Statement II is false.

(D) Statement I is incorrect but Statement II is true.

62. Intramolecular hydrogen bonding is present in:



(D) HF

63. On heating, some solid substances change from solid to vapour state without passing through liquid state. The technique used for the purification of such solid substances based on the above principle is known as:

(A) Crystallization

(B) Sublimation

(C) Distillation

(D) Chromatography

64. In which of the following processes entropy increases?

a. A liquid evaporates to vapour

b. Temperature of a crystalline solid lowered from 130 K to 0 K.

c. $2\text{NaHCO}_3(s) \rightarrow \text{Na}_2\text{CO}_3(s) + \text{CO}_2(g) + \text{H}_2\text{O}(g)$

d. $\text{Cl}_2(g) \rightarrow 2\text{Cl}(g)$

Choose the correct answer from the options given below:

(A) a and c

(B) a, b and d

(C) a, c and d

(D) c and d

65. Among Group 16 elements, which one does NOT show -2 oxidation state?

(A) O

(B) Se

(C) Te

(D) Po

66. Match List-I with List-II.

List-I

(Conversion)

(A) 1 mol of H_2O to O_2

(B) 1 mol of MnO_4^- to Mn^{2+}

(C) 1.5 mole of Ca from molten CaCl_2

(D) 1 mol of FeO to Fe_2O_3

(A) A-II, B-IV, C-I, D-III

(C) A-II, B-III, C-I, D-IV

List-II

(Number of faraday required)

(I) 3 F

(II) 2 F

(III) 1 F

(IV) 5 F

(B) A-III, B-IV, C-I, D-II

(D) A-III, B-IV, C-II, D-I

67. Arrange the following elements in increasing order of electronegativity.

N, O, F, C, Si

Choose the correct answer from the options given below:

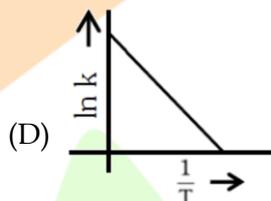
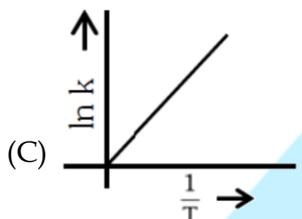
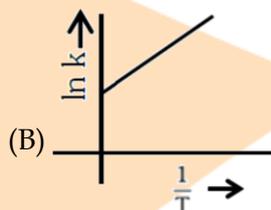
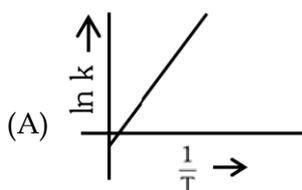
(A) $\text{Si} < \text{C} < \text{N} < \text{O} < \text{F}$

(B) $\text{Si} < \text{C} < \text{O} < \text{N} < \text{F}$

(C) $\text{O} < \text{F} < \text{N} < \text{C} < \text{Si}$

(D) $\text{F} < \text{O} < \text{N} < \text{C} < \text{Si}$

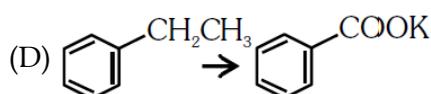
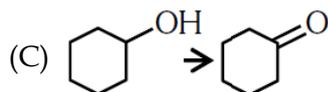
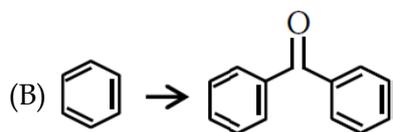
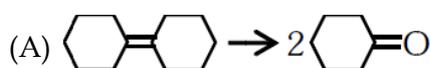
68. A compound with a molecular formula of C_6H_{14} has two tertiary carbons. Its IUPAC name is :
 (A) n-hexane (B) 2-methylpentane
 (C) 2, 3-dimethylbutane (D) 2, 2-dimethylbutane
69. Fehling's solution 'A' is
 (A) aqueous copper sulphate
 (B) alkaline copper sulphate
 (C) alkaline solution of sodium potassium tartrate (Rochelle's salt)
 (D) aqueous sodium citrate
70. Activation energy of any chemical reaction can be calculated if one knows the value of
 (A) rate constant at standard temperature.
 (B) probability of collision.
 (C) orientation of reactant molecules during collision.
 (D) rate constant at two different temperatures.
71. Which plot of $\ln k$ vs $\frac{1}{T}$ is consistent with Arrhenius equation?



72. Match List I with List II.

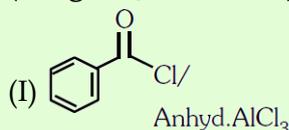
List I

(Reaction)



List II

(Reagents/Condition)



(II) CrO_3

(III) $KMnO_4/KOH, \Delta$

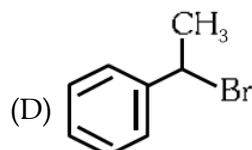
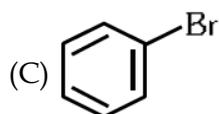
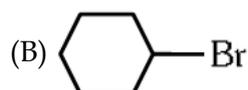
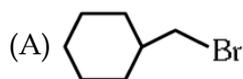
(IV) (i) O_3 (ii) $Zn-H_2O$

Choose the correct answer from the options given below:

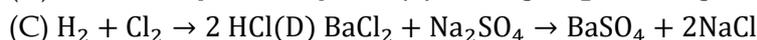
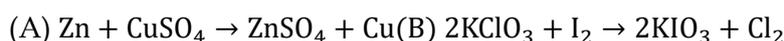
- (A) A-IV, B-I, C-III, D-II
 (C) A-IV, B-I, C-II, D-III

- (B) A-III, B-I, C-II, D-IV
 (D) A-I, B-IV, C-II, D-III

73. The compound that will undergo S_N1 reaction with the fastest rate is:



74. Which reaction is NOT a redox reaction?



75. Given below are two statements :

Statement I : The boiling point of three isomeric pentanes follows the order n-pentane > isopentane > neopentane

Statement II : When branching increases, the molecule attains a shape of sphere.

This results in smaller surface area for contact, due to which the intermolecular forces between the spherical molecules are weak, thereby lowering the boiling point.

In the light of the above statements, choose the most appropriate answer from the options given below:

- (A) Both statement I and Statement II are correct
 (B) Both Statement I and Statement II are incorrect
 (C) Statement I is correct but Statement II is incorrect
 (D) Statement I is incorrect but Statement II is correct

76. Given below are two statements:

Statement I : Both $[Co(NH_3)_6]^{+3}$ and $[CoF_6]^{3-}$ complexes are octahedral but differ in their magnetic behaviour.

Statement II : $[Co(NH_3)_6]^{3+}$ is diamagnetic whereas $[CoF_6]^{3-}$ is paramagnetic.

In the light of the above statements, choose the correct answer from the options given below :

- (A) Both statement I and Statement II are true
 (B) Both Statement I and Statement II are false
 (C) Statement I is true but Statement II is false
 (D) Statement I is false but Statement II is true

77. Match List I with List II.

List-I (Molecule)		List-II (Number and types of bond/s between two carbon atoms)	
A.	ethane	I.	One σ -bond and two π -bond
B.	ethene	II.	Two π -bonds
C.	Carbon molecule, C_2	III.	One σ -bond
D.	ethyne	IV.	One π -bonds and one π -bond

Chose the correct answer from the option given below

(A) A-I, B-IV, C-II, D-III

(B) A-IV, B-III, C-II, D-I

(C) A-III, B-IV, C-II, D-I

(D) A-III, B-IV, C-I, D-II

78. The Henry's law constant (K_H) values of three gases (A, B, C) in water are 145, 2×10^{-5} and 35 kbar, respectively. The solubility of these gases in water follow the order:
 (A) $B > A > C$ (B) $B > C > A$
 (C) $A > C > B$ (D) $A > B > C$
79. The energy of an electron in the ground state ($n = 1$) for He^+ ion is $-x\text{J}$. Then that for an electron in $n = 2$ state for Be^{3+} ion in J is:
 (A) $-x$ (B) $-\frac{x}{9}$ (C) $-4x$ (D) $-\frac{4}{9}x$
80. The Σ° value for the $\text{Mn}^{3+}/\text{Mn}^{2+}$ couple is more positive than that of $\text{Cr}^{3+}/\text{Cr}^{2+}$ or $\text{Fe}^{3+}/\text{Fe}^{2+}$ due to change of
 (A) d^5 to d^4 configuration (B) d^5 to d^2 configuration
 (C) d^4 to d^5 configuration (D) d^3 to d^5 configuration
81. The reagents with which glucose does not react to give the corresponding tests/products are
 A. Tollens' reagent B. Schiff's reagent
 C. HCN D. NH_2OH
 E. NaHSO_3
 Choose the correct options from the given below:
 (A) B and C (B) A and D
 (C) B and E (D) E and D

82. Match List I with List II.

List-I (Complex)		List-II (Type of isomerism)	
A.	$[\text{Co}(\text{NH}_3)_5(\text{NO}_2)]\text{Cl}_2$	I.	Solvate isomerism
B.	$[\text{Co}(\text{NH}_3)_5(\text{SO}_4)]\text{Br}$	II.	Linkage isomerism
C.	$[\text{Co}(\text{NH}_3)_6][\text{Cr}(\text{CN})_6]$	III.	Ionization isomerism
D.	$[\text{Co}(\text{H}_2\text{O})_6]\text{Cl}_3$	IV.	Coordination isomerism

Choose the correct answer from the options given below:

- (A) A-II, B-III, C-IV, D-I (B) A-I, B-III, C-IV, D-II
 (C) A-I, B-IV, C-III, D-II (D) A-II, B-IV, C-III, D-I
83. Arrange the following elements in increasing order of first ionization enthalpy :
 Li, Be, B, C, N
 Chose the correct answer from the options given below:
 (A) $\text{Li} < \text{Be} < \text{B} < \text{C} < \text{N}$ (B) $\text{Li} < \text{B} < \text{Be} < \text{C} < \text{N}$
 (C) $\text{Li} < \text{Be} < \text{C} < \text{B} < \text{N}$ (D) $\text{Li} < \text{Be} < \text{N} < \text{B} < \text{C}$

84. 1 gram of sodium hydroxide was treated with 25 mL of 0.75 M HCl solution, the mass of sodium hydroxide left unreacted is equal to
(A) 750 mg (B) 250 mg (C) Zero mg (D) 200 mg

85. For the reaction $2A \rightleftharpoons B + C$, $K_c = 4 \times 10^{-3}$. At a given time, the composition of reaction mixture is:

$$[A] = [B] = [C] = 2 \times 10^{-3} M$$

Then, which of the following is correct?

- (A) Reaction is at equilibrium.
(B) Reaction has a tendency to go in forward direction.
(C) Reaction has a tendency to go in backward direction
(D) Reaction has gone to completion in forward direction.

Section - B

86. Given below are certain cations. Using inorganic qualitative analysis, arrange them in increasing group number from 0 to VI.

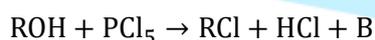
A. Al^{3+} B. Cu^{2+} C. Ba^{2+} D. Co^{2+}

E. Mg^{2+}

Choose the correct answer from the options given below:

- (A) B, A, D, C, E (B) B, C, A, D, E
(C) E, C, D, B, A (D) E, A, B, C, D

87. The products A and B obtained in the following reactions, respectively, are



- (A) $POCl_3$ and H_3PO_3 (B) $POCl_3$ and I_3PO_4
(C) H_3PO_4 and $POCl_3$ (D) H_3PO_3 and $POCl_3$

88. Mass in grams of copper deposited by passing 9.6487 A current through a voltmeter containing copper sulphate solution for 100 seconds is:

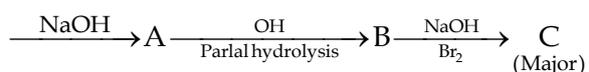
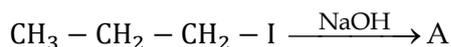
(Given : Molar mass of Cu: 63 g mol^{-1} , $1 F = 96487 C$)

- (A) 3.15 g (B) 0.315 g (C) 31.5 g (D) 0.0315 g

89. The plot of osmotic pressure (Π) vs concentration (mol L^{-1}) for a solution gives a straight line with slope $25.73 \text{ L}^{-1} \text{ bar mol}^{-1}$. The temperature at which the osmotic pressure measurement is done is : (Use $R = 0.083 \text{ L}^{-1} \text{ bar mol}^{-1} \text{ K}^{-1}$)

- (A) $37^\circ C$ (B) $310^\circ C$ (C) $25.73^\circ C$ (D) $12.05^\circ C$

90. Identify the major product *C* formed in the following reaction sequence:



- (A) propylamine (B) Butylamine
(C) butanamide (D) α - promobutanoic acid

91. Identify the correct answer.

- (A) Three resonance structures can be drawn for ozone
(B) BF_3 has non-zero dipole moment
(C) Dipole moment of NF_3 is greater than that of NH_3
(D) Three canonical forms can be drawn for CO_3^{2-} ion.

92. Given below are two statements:

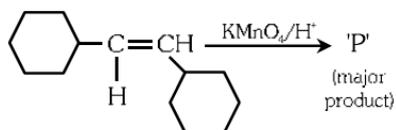
Statement I : $[\text{Co}(\text{NH}_3)_6]^{3+}$ is a homoleptic complex whereas $[\text{Co}(\text{NH}_3)_4\text{Cl}_2]^+$ is a heteroleptic complex.

Statement II : Complex $[\text{Co}(\text{NH}_3)_6]^{3+}$ has only one kind of ligands but $[\text{Co}(\text{NH}_3)_4\text{Cl}_2]^+$ has more than one kind of ligands.

In the light of the above statements, choose the correct answer from the options given below.

- (A) Both Statement I and Statement II are true.
(B) Both Statement I and Statement II are false.
(C) Statement I is true but Statement II is false.
(D) Statement I is false but Statement II is true.

93. For the given reaction



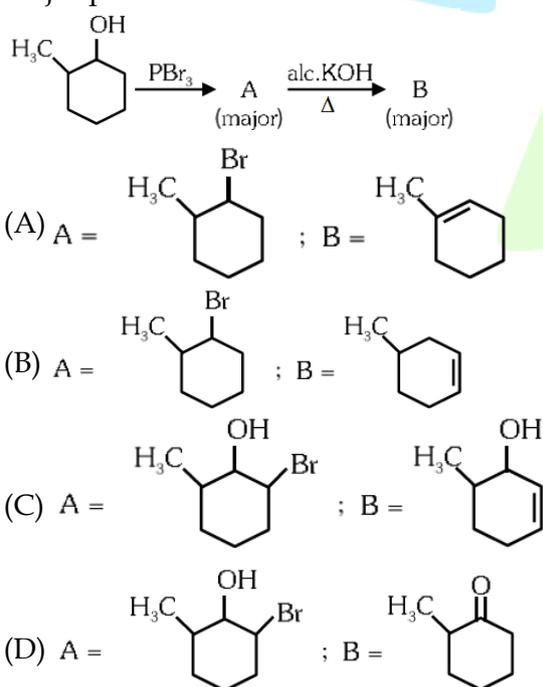
'P' is

- (A) (B)
(C) (D)

94. The pair of lanthanoid ions which are diamagnetic is

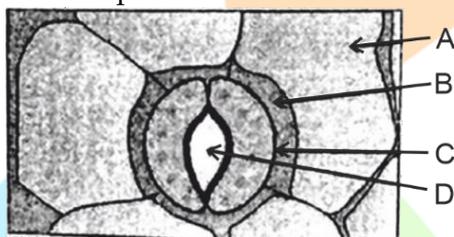
- (A) Ce^{4+} and Yb^{2+} (B) Ce^{3+} and Eu^{2+}
(C) Gd^{3+} and Eu^{3+} (D) Pm^{3+} and Sm^{3+}

95. Consider the following reaction in a sealed vessel at equilibrium with concentrations of $N_2 = 3.0 \times 10^{-3}M$, $O_2 = 4.2 \times 10^{-3}M$ and $NO = 2.8 \times 10^{-3}M$.
 $2NO_{(g)} \rightleftharpoons N_{2(g)} + O_{2(g)}$
 If 0.1 mol L^{-1} of $NO_{(g)}$ is taken in a closed vessel, what will be degree of dissociation (α) of $NO_{(g)}$, at equilibrium?
 (A) 0.00889 (B) 0.0889
 (C) 0.8889 (D) 0.717
96. A compound X contains 32% of A, 20% of B and remaining percentage of C. Then, the empirical formula of X is :
 (Given atomic masses of A = 64; B = 40; C = 32u)
 (A) A_2BC_2 (B) ABC_3 (C) AB_2C_2 (D) ABC_4
97. The work done during reversible isothermal expansion of one mole of hydrogen gas at $25^\circ C$ from pressure of 20 atmosphere to 10 atmosphere is :
 (Given $R = 2.0 \text{ cal K}^{-1} \text{ mol}^{-1}$)
 (A) 0 calorie (B) -413.14 calories (C) 413.14 calories (D) 100 calories
98. During the preparation of Mohr's salt solution (Ferrous ammonium sulphate), Which of the following acid is added to prevent hydrolysis of Fe^{2+} ion?
 (A) dilute hydrochloric acid
 (B) concentrated sulphuric acid
 (C) dilute nitric acid
 (D) dilute sulphuric acid
99. The rate of a reaction quadruples when temperature changes from $27^\circ C$ to $57^\circ C$. Calculate the energy of activation.
 Given $R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$, $\log 4 = 0.6021$
 (A) $\frac{38.04 \text{ kJ}}{\text{mol}}$ (B) $\frac{380.4 \text{ kJ}}{\text{mol}}$ (C) $\frac{3.80 \text{ kJ}}{\text{mol}}$ (D) 3804 kJ/mol
100. Major products A and B formed in the following reaction sequence, are



101. Lecithin, a small molecular weight organic compound found in living tissues, is an example of:
 (A) Amino acids (B) Phospholipids
 (C) Glycerides (D) Carbohydrates
102. Which of the following are required for the dark reaction of photosynthesis ?
 A. Light B. Chlorophyll
 C. CO₂ D. ATP
 E. NADPH
 Choose the **correct** answer from the options given below:
 (A) A, B and C only (B) B, C and D only
 (C) C, D and E only (D) D and E only
103. Spindle fibers attach to kinetochores of chromosomes during
 (A) Prophase (B) Metaphase (C) Anaphase (D) Telophase
104. Bulliform cells are responsible for
 (A) Inward curling of leaves in monocots.
 (B) Protecting the plant from salt stress.
 (C) Increased photosynthesis in monocots.
 (D) Providing large spaces for storage of sugars.

105. In the given figure, which component has thin outer walls and highly thickened inner walls?

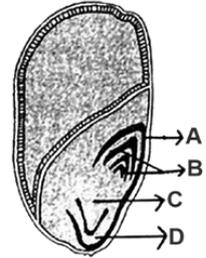


- (A) C (B) D (C) A (D) B
106. What is the fate of a piece of DNA carrying only gene of interest which is transferred into an alien organism?
 A. The piece of DNA would be able to multiply itself independently in the progeny cells of the organism.
 B. It may get integrated into the genome of the recipient.
 C. It may multiply and be inherited along with the host DNA.
 D. The alien piece of DNA is not an integral part of chromosome.
 E. It shows ability to replicate.
 Choose the correct answer from the options given below:
 (A) A and B only (B) D and E only
 (C) B and C only (D) A and E only
107. Given below are two statements:
Statement I : Bt toxins are insect group specific and coded by a gene *cry* IAc.
Statement II : Bt toxin exists as inactive protoxin in *B. thuringiensis*. However, after ingestion by the insect the inactive protoxin gets converted into active form due to acidic pH of the insect gut.
 In the light of the above statements, choose the correct answer from the options given below:
 (A) Both Statement I and Statement II are true
 (B) Both Statement I and Statement II are false
 (C) Statement I is true but Statement II is false
 (D) Statement I is false but Statement II is true

108. List of endangered species was released by
 (A) GEAC (B) WWF (C) FOAM (D) IUCN

109. Identify the part of the seed from the given figure which is destined to form root when the seed germinates.

- (A) A (B) B (C) C (D) D



110. Match List I with List II

List I

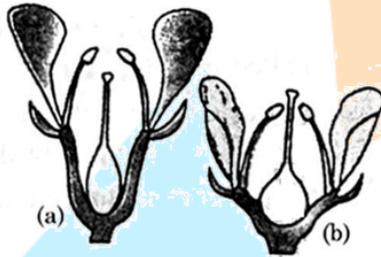
- A. *Clostridium butylicum*
 B. *Saccharomyces cerevisiae*
 C. *Trichoderma polysporum*
 D. *Streptococcus sp.* IV. Cyclosporin-A

List II

- I. Ethanol
 II. Streptokinase
 III. Butyric acid

Choose the correct answer from the options given below:

- (A) A-III, B-I, C-II, D-IV (B) A-II, B-IV, C-III, D-I
 (C) A-III, B-I, C-IV, D-II (D) A-IV, B-I, C-III, D-II
111. Identify the type of flowers based on the position of calyx, corolla and androecium with respect to the ovary from the given figures (a) and (b)



- (A) (a) Epigynous; (b) Hypogynous (B) (a) Hypogynous; (b) Epigynous
 (C) (a) Perigynous; (b) Epigynous (D) (a) Perigynous; (b) Perigynous

112. Auxin is used by gardeners to prepare weed-free lawns. But no damage is caused to grass as auxin
 (A) promotes apical dominance.

- (B) promotes abscission of mature leaves only.
 (C) does not affect mature monocotyledonous plants.
 (D) can help in cell division in grasses, to produce growth.

113. A pink flowered Snapdragon plant was crossed with a red flowered Snapdragon plant. What type of phenotype/s is/are expected in the progeny?

- (A) Only red flowered plants
 (B) Red flowered as well as pink flowered plants
 (C) Only pink flowered plants
 (D) Red, Pink as well as white flowered plants

114. Which one of the following is not a criterion for classification of fungi?

- (A) Morphology of mycelium (B) Mode of nutrition
 (C) Mode of spore formation (D) Fruiting body

115. The lactose present in the growth medium of bacteria is transported to the cell by the action of
(A) Beta-galactosidase (B) Acetylase (C) Permease (D) Polymerase
116. In a plant, black seed color (BB/Bb) is dominant over white seed color (bb). In order to find out the genotype of the black seed plant, with which of the following genotype will you cross it?
(A) BB (B) bb (C) Bb (D) BB/Bb
117. Given below are two statements:
Statement I : Parenchyma is living but collenchyma is dead tissue.
Statement II : Gymnosperms lack xylem vessels but presence of xylem vessels is the characteristic of angiosperms.
In the light of the above statements, choose the correct answer from the options given below:
(A) Both Statement I and Statement II are true
(B) Both Statement I and Statement II are false
(C) Statement I is true but Statement II is false
(D) Statement I is false but Statement II is true
118. How many molecules of ATP and NADPH are required for every molecule of CO₂ fixed in the Calvin cycle?
(A) 2 molecules of ATP and 3 molecules of NADPH
(B) 2 molecules of ATP and 2 molecules of NADPH
(C) 3 molecules of ATP and 3 molecules of NADPH
(D) 3 molecules of ATP and 2 molecules of NADPH
119. A transcription unit in DNA is defined primarily by the three regions in DNA and these are with respect to upstream and down stream end;
(A) Repressor, Operator gene, Structural gene (B) Structural gene, Transposons, Operator gene
(C) Inducer, Repressor, Structural gene (D) Promotor, Structural gene, Terminator
120. Tropical regions show greatest level of species richness because
A. Tropical latitudes have remained relatively undisturbed for millions of years, hence more time was available for species diversification.
B. Tropical environments are more seasonal.
C. More solar energy is available in tropics.
D. Constant environments promote niche specialization.
E. Tropical environments are constant and predictable. Choose the correct answer from the options given below.
(A) A, C, D and E only (B) A and B only
(C) A, B and E only (D) A, B and D only
121. The equation of Verhulst-Pearl logistic growth is $\frac{dN}{dt} = rN \left[\frac{K-N}{K} \right]$.
From this equation, K indicates:
(A) Intrinsic rate of natural increase (B) Biotic potential
(C) Carrying capacity (D) Population density
122. Inhibition of Succinic dehydrogenase enzyme by malonate is a classical example of:
(A) Cofactor inhibition (B) Feedback inhibition
(C) Competitive inhibition (D) Enzyme activation

126. Match List I with List II

	List-I		List-II
A.	<i>Rhizopus</i>	I.	Mushroom
B.	<i>Ustilago</i>	II.	Smut fungus
C.	<i>Puccinia</i>	III.	Bread mould
D.	<i>Agaricus</i>	IV.	Rust fungus

Choose the correct answer from the options given below:

(A) A-III, B-II, C-IV, D-I

(B) A-I, B-III, C-II, D-IV

(C) A-III, B-II, C-I, D-IV

(D) A-IV, B-III, C-II, D-I

127. Hind II always cuts DNA molecules at a particular point called recognition sequence and it consists of :

(A) 8 bp

(B) 6 bp

(C) 4 bp

(D) 10 bp

128. Which of the following is an example of actinomorphic flower ?

(A) *Datura*

(B) *Cassia*

(C) *Pisum*

(D) *Sesbania*

129. The type of conservation in which the threatened species are taken out from their natural habitat and placed in special setting where they can be protected and given special care is called

(A) *in-situ* conservation

(B) Biodiversity conservation

(C) Semi-conservative method

(D) Sustainable development

130. Given below are two statements:

Statement I : Chromosomes become gradually visible under light microscope during leptotene stage.

Statement II : The beginning of diplotene stage is recognized by dissolution of synaptonemal complex. In the light of the above statements, choose the correct answer from the options given below:

(A) Both Statement I and Statement II are true

(B) Both Statement I and Statement II are false

(C) Statement I is true but Statement II is false

(D) Statement I is false but Statement II is true

131. Formation of interfascicular cambium from fully developed parenchyma cells is an example for

(A) Differentiation

(B) Redifferentiation

(C) Dedifferentiation

(D) Maturation

132. The capacity to generate a whole plant from any cell of the plant is called:

(A) Totipotency

(B) Micropropagation

(C) Differentiation

(D) Somatic hybridization

133. Match List I with List II

List I

- A. Two or more alternative forms of a gene
- B. Cross of F_1 progeny with homozygous
- C. Cross of F_1 progeny with any of the parents
- D. Number of chromosome sets in plant

Choose the **correct** answer from the options given below:

- (A) A-I, B-II, C-III, D-IV
- (B) A-II, B-I, C-III, D-IV
- (C) A-III, B-IV, C-I, D-II
- (D) A-IV, B-III, C-II, D-I

List II

- I. Back cross
- II. Ploidy recessive parent
- III. Allele
- IV. Test cross

134. The cofactor of the enzyme carboxypeptidase is:

- (A) Zinc
- (B) Niacin
- (C) Flavin
- (D) Haem

135. These are regarded as major causes of biodiversity loss:

- A. Over exploitation
- B. Co-extinction
- C. Mutation
- D. Habitat loss and fragmentation
- E. Migration

Choose the correct option:

- (A) A, C and D only
- (B) A, B, C and D only
- (C) A, B and E only
- (D) A, B and D only

SECTION-B

136. Match List I with List II

List I (Types of Stamens)

- A. Monoadelphous
 B. Diadelphous
 C. Polyadelphous
 D. Epiphyllous

List II (Example)

- I. Citrus
 II. Pea
 III. Lily
 IV. China-rose

Choose the correct answer from the options given below:

- (A) A-IV, B-II, C-I, D-III
 (B) A-IV, B-I, C-II, D-III
 (C) A-I, B-II, C-IV, D-III
 (D) A-III, B-I, C-IV, D-II

137. Match List-I with List-II

List-I

- A. GLUT-4
 B. Insulin
 C. Trypsin
 D. Collagen

List-II

- I. Hormone
 II. Enzyme
 III. Intercellular ground substance
 IV. Enables glucose transport into cells

Choose the correct answer from the options given below.

- (A) A-IV, B-I, C-II, D-III
 (B) A-I, B-II, C-III, D-IV
 (C) A-II, B-III, C-IV, D-I
 (D) A-III, B-IV, C-I, D-II

138. Identify the step in tricarboxylic acid cycle, which does not involve oxidation of substrate.

- (A) Malic acid \rightarrow Oxaloacetic acid
 (B) Succinic acid \rightarrow Malic acid
 (C) Succinyl-CoA \rightarrow Succinic acid
 (D) Isocitrate \rightarrow α -ketoglutaric acid

139. Match List I with List II

	List I		List II
A.	Citric acid cycle	I.	Cytoplasm
B.	Glycolysis	II.	Mitochondrial matrix
C.	Electron transport system	III.	Intermembrane space of mitochondria
D.	Proton gradient	IV.	Inner mitochondrial membrane

Choose the correct answer from the options given below:

- (A) A-I, B-II, C-III, D-IV
 (B) A-II, B-I, C-IV, D-III
 (C) A-III, B-IV, C-I, D-II
 (D) A-IV, B-III, C-II, D-I

140. Match List I with List II

List I

- A. Frederick Griffith I. Genetic code
 B. Francois Jacob & Jacque Monod
 C. Har Gobind Khorana
 D. Meselson & Stahl IV. Lac operon

List II

- II. Semi-conservative mode of DNA replication
 III. Transformation

Choose the correct answer from the options given below:

- (A) A-III, B-II, C-I, D-IV
 (B) A-III, B-IV, C-I, D-II
 (C) A-II, B-III, C-IV, D-I
 (D) A-IV, B-I, C-II, D-III

141. Given below are two statements:

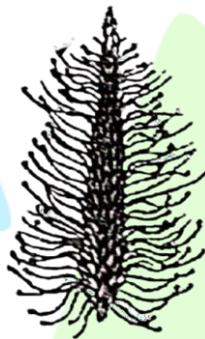
Statement I: In C_3 plants, some O_2 binds to RuBisCO, hence CO_2 fixation is decreased.

Statement II: In C_4 plants, mesophyll cells show very little photorespiration while bundle sheath cells do not show photorespiration.

In the light of the above statements, choose the *correct* answer from the options given below:

- (A) Both Statement I and Statement II are true
 (B) Both Statement I and Statement II are false
 (C) Statement I is true but Statement II is false
 (D) Statement I is false but Statement II is true

142. Identify the correct description about the given figure:



- (A) Wind pollinated plant inflorescence showing flowers with well exposed stamens.
 (B) Water pollinated flowers showing stamens with mucilaginous covering.
 (C) Cleistogamous flowers showing autogamy.
 (D) Compact inflorescence showing complete autogamy

143. Match List I with List II

List I

- A. Rose
 B. Pea
 C. Cotton
 D. Mango

List II

- I. Twisted aestivation
 II. Perigynous flower
 III. Drupe
 IV. Marginal placentation

Choose the correct answer from the options given below :

- (A) A-II, B-IV, C-I, D-III
 (B) A-I, B-II, C-III, D-IV
 (C) A-IV, B-III, C-II, D-I
 (D) A-II, B-III, C-IV, D-I

144. Read the following statements and choose the set of correct statements: In the members of Phaeophyceae,
- A. Asexual reproduction occurs usually by biflagellate zoospores.
 - B. Sexual reproduction is by oogamous method only.
 - C. Stored food is in the form of carbohydrates which is either mannitol or laminarin.
 - D. The major pigments found are chlorophyll a, c and carotenoids and xanthophyll.
 - E. Vegetative cells have a cellulosic wall, usually covered on the outside by gelatinous coating of algin.
- Choose the correct answer from the options given below:
- (A) A, B, C and D only
 - (B) B, C, D and E only
 - (C) A, C, D and E only
 - (D) A, B, C and E only
145. In an ecosystem if the Net Primary Productivity (NPP) of first trophic level is $100x \text{ (kcal m}^{-2}\text{) yr}^{-1}$, what would be the GPP (Gross Primary Productivity) of the third trophic level of the same ecosystem?
- (A) $\frac{x}{10} \text{ (kcal m}^{-2}\text{) yr}^{-1}$
 - (B) $x \text{ (kcal m}^{-2}\text{) yr}^{-1}$
 - (C) $10x \text{ (kcal m}^{-2}\text{) yr}^{-1}$
 - (D) $\frac{100x}{3x} \text{ (kcal m}^{-2}\text{) yr}^{-1}$
146. Which of the following statement is correct regarding the process of replication in *E.coli*?
- (A) The DNA dependent DNA polymerase catalyses polymerization in one direction that is $3' \rightarrow 5'$
 - (B) The DNA dependent RNA polymerase catalyses polymerization in one direction, that is $5' \rightarrow 3'$
 - (C) The DNA dependent DNA polymerase catalyses polymerization in $5' \rightarrow 3'$ as well as $3' \rightarrow 5'$ direction
 - (D) The DNA dependent DNA polymerase catalyses polymerization in $5' \rightarrow 3'$ direction
147. Which of the following are fused in somatic hybridization involving two varieties of plants?
- (A) Callus
 - (B) Somatic embryos
 - (C) Protoplasts
 - (D) Pollens
148. Spraying sugarcane crop with which of the following plant growth regulators, increases the length of stem, thus, increasing the yield?
- (A) Auxin
 - (B) Gibberellin
 - (C) Cytokinin
 - (D) Abscisic acid

149. Match List I with List II

List I

- A. Robert May
B. Alexander von Humboldt
C. Paul Ehrlich
D. David Tilman

List II

- I. Species-Area relationship
II. Long term ecosystem experiment using out door plots
III. Global species diversity at about 7 million
IV. Rivet popper hypothesis

Choose the correct answer from the options given below:

- (A) A-II, B-III, C-I, D-IV
(B) A-III, B-I, C-IV, D-II
(C) A-I, B-III, C-II, D-IV
(D) A-III, B-IV, C-II, D-I

150. The DNA present in chloroplast is:

- (A) Linear, double stranded
(B) Circular, double stranded
(C) Linear, single stranded
(D) Circular, single stranded

151. Match List I with List II :

	List I		List II
A.	Common cold	I.	Plasmodium
B.	Haemozoin	II.	Typhoid
C.	Widal test	III.	Rhinoviruses
D.	Allergy	IV.	Dust mites

Choose the correct answer from the options given below :

- (A) A-II, B-IV, C-III, D-I
(B) A-I, B-III, C-II, D-IV
(C) A-III, B-I, C-II, D-IV
(D) A-IV, B-II, C-III, D-I

152. Match List I with List II :

	List I		List II
A.	Cocaine	I.	Effective sedative in surgery
B.	Heroin	II .	Cannabis sativa
C.	Morphine	III.	Erythroxyllum
D.	Marijuana	IV.	Papaver somniferum

Choose the correct answer from the options given below:

- (A) A-IV, B-III, C-I, D-II
(B) A-I, B-III, C-II, D-IV
(C) A-II, B-I, C-III, D-IV
(D) A-III, B-IV, C-I, D-II

153. Match List I with List II :

List I

- A. Fibrous joints
B. Cartilaginous joints
C. Hinge joints
D. Ball and socket joints

List II

- I. Adjacent vertebrae, limited movement
II. Humerus and Pectoral girdle, rotational movement
III. Skull, don't allow any movement
IV. Knee, help in locomotion

Choose the correct answer from the options given below :

- (A) A-IV, B-II, C-III, D-I
(B) A-I, B-III, C-II, D-IV
(C) A-II, B-III, C-I, D-IV
(D) A-III, B-I, C-IV, D-II

154. Which of the following are Autoimmune disorders?

- A. Myasthenia gravis B. Rheumatoid arthritis
C. Gout D. Muscular dystrophy
E. Systemic Lupus Erythematosus (SLE)

Choose the most appropriate answer from the options given below:

- (A) A, B & D only (B) A, B & E only
(C) B, C & E only (D) C, D & E only

155. Which of the following is not a component of Fallopian tube?

- (A) Uterine fundus (B) Isthmus
(C) Infundibulum (D) Ampulla

156. The flippers of the Penguins and Dolphins are the example of the

- (A) Adaptive radiation (B) Natural selection
(C) Convergent evolution (D) Divergent evolution

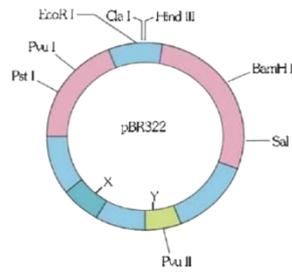
157. Match List I with List II:

	List I		List II
A.	α -1 antitrypsin	I.	Cotton bollworm
B.	Cry IAb	II.	ADA deliciency
C.	Cry IAc	III.	Emphysema
D.	Enzyme replacement therapy	IV.	Corn borer

Choose the correct answer from the options given below:

- (A) A-II, B-I, C-IV, D-III
(B) A-III, B-I, C-II, D-IV
(C) A-III, B-IV, C-I, D-II
(D) A-II, B-IV, C-I, D-III

158. The following diagram showing restriction sites in *E. coli* cloning vector pBR322. Find the role of 'X' and 'Y' genes.



- (A) The gene 'X' is responsible for resistance to antibiotics and 'Y' for protein involved in the replication of Plasmid.
- (B) The gene 'X' is responsible for controlling the copy number of the linked DNA and 'Y' for protein involved in the replication of Plasmid.
- (C) The gene 'X' is for protein involved in replication of Plasmid and 'Y' for resistance to antibiotics.
- (D) Gene 'X' is responsible for recognition sites and 'Y' is responsible for antibiotic resistance
159. Given below are two statements: One is labelled as Assertion A and the other is labelled as Reason R:
Assertion A : Breast-feeding during initial period of infant growth is recommended by doctors for bringing a healthy baby.
Reason R : Colostrum contains several antibodies absolutely essential to develop resistance for the new born baby.
 In the light of the above statements, choose the most appropriate answer from the options given below:
- (A) Both A and R are correct and R is the correct explanation of A
 (B) Both A and R are correct but R is NOT the correct explanation of A
 (C) A is correct but R is not correct
 (D) A is not correct but R is correct

160. The "Ti plasmid" of *Agrobacterium tumefaciens* stands for
- (A) Tumour inhibiting plasmid (B) Tumor independent plasmid
 (C) Tumor inducing plasmid (D) Temperature independent plasmid
161. Match List I with List II :

	List I		List II
A.	Pleurobrachia	I.	Mollusca
B.	Radula	II.	Ctenophora
C.	Stomochord	III.	Osteichthyes
D.	Air bladder	IV.	Hemichordata

Choose the correct answer from the options given below :

- (A) A-IV, B-II, C-III, D-I
 (B) A-II, B-I, C-IV, D-III
 (C) A-II, B-IV, C-I, D-III
 (D) A-IV, B-III, C-II, D-I

162. Given below are some stages of human evolution. Arrange them in correct sequence. (Past to Recent)

- A. Homo habilis
- B. Homo sapiens
- C. Homo neanderthalensis
- D. Homo erectus

Choose the correct sequence of human evolution from the options given below:

- (A) D-A-C-B
- (B) B-A-D-C
- (C) C-B-D-A
- (D) A-D-C-B

163. Which of the following is not a steroid hormone?

- (A) Cortisol
- (B) Testosterone
- (C) Progesterone
- (D) Glucagon

164. In both sexes of cockroach, a pair of jointed filamentous structures called anal cerci are present on

- (A) 5th segment
- (B) 10th segment
- (C) 8th and 9th segment
- (D) 11th segment

165. Which one of the following factors will not affect the Hardy-Weinberg equilibrium?

- (A) Genetic recombination
- (B) Genetic drift
- (C) Gene migration
- (D) Constant gene pool

166. Match List I with List II :

	List I		List II
A.	Pons	I.	Provides additional space for Neurons, regulates posture and balance.
B.	Hypothalamus	II.	Controls respiration and gastric secretions.
C.	Medulla	III.	Connects different regions of the brain.
D.	Cerebellum	IV.	Neuro secretory cells

Choose the correct answer from the options given below :

- (A) A-II, B-III, C-I, D-IV
- (B) A-III, B-IV, C-II, D-I
- (C) A-I, B-III, C-II, D-IV
- (D) A-II, B-I, C-III, D-IV

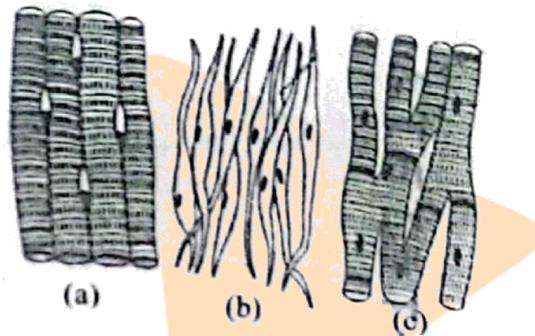
172. Consider the following statements :

- A. Annelids are true coelomates
- B. Poriferans are pseudocoelomates
- C. Aschelminthes are acoelomates
- D. Platyhelminthes are pseudocoelomates

Choose the correct answer from the options given below :

- (A) B only
- (B) A only
- (C) C only
- (D) D only

173. Three types of muscles are given as a, b and c. Identify the correct matching pair along with their location in human body:



Name of muscle/location

- (A) (a) Smooth - Toes (b) Skeletal - Legs (c) Cardiac - Heart
- (B) (a) Skeletal - Triceps (b) Smooth - Stomach (c) Cardiac - Heart
- (C) (a) Skeletal - Biceps (b) Involuntary - Intestine (c) Smooth - Heart
- (D) (a) Involuntary - Nose tip (b) Skeletal - Bone (c) Cardiac - Heart

174. Following are the stages of pathway for conduction of an action potential through the heart

- A. AV bundle
- B. Purkinje fibres
- C. AV node
- D. Bundle branches
- E. SA node

Choose the correct sequence of pathway from the options given below

- (A) E-C-A-D-B
- (B) A-E-C-B-D
- (C) B-D-E-C-A
- (D) E-A-D-B-C

175. Match List I with List II :

	List-I		List-II
A.	Lipase	I.	Peptide bond
B.	Nuclease	II.	Ester bond
C.	Protease	III.	Glycosidic bond
D.	Amylase	IV.	Phosphodiester bond

Choose the correct answer from the options given below :

- (A) A-IV, B-II, C-III, D-I
- (B) A-III, B-II, C-I, D-IV
- (C) A-II, B-IV, C-I, D-III
- (D) A-IV, B-I, C-III, D-II

176. Match List I with List II :

List I

- A. Axoneme
B. Cartwheel pattern
C. Crista
D. Satellite

List II

- I. Centriole
II. Cilia and flagella
III. Chromosome
IV. Mitochondria

Choose the correct answer from the options given below :

- (A) A-IV, B-III, C-II, D-I
(B) A-IV, B-II, C-III, D-I
(C) A-II, B-IV, C-I, D-III
(D) A-II, B-I, C-IV, D-III

177. Match List I with List II :

	List I (Sub Phases of Prophase I)		List II (Specific Characters)
A.	Diakinesis	I.	Synaptonemal complex formation
B.	Pachytene	II.	Completion of terminalisation of chiasmata
C.	Zygotene	III.	Chromosomes look like thin threads
D.	Leptotene	IV.	Appearance of recombination nodules

Choose the correct answer from the options given below

- (A) A-IV, B-II, C-III, D-I
(B) A-I, B-II, C-IV, D-III
(C) A-II, B-IV, C-I, D-III
(D) A-IV, B-III, C-II, D-I

178. Which of the following factors are favourable for the formation of oxyhaemoglobin in alveoli?

- (A) High pO_2 and High pCO_2
(B) High pO_2 and Lesser H^+ concentration
(C) Low pCO_2 and High H^+ concentration
(D) Low pCO_2 and High temperature

179. Match List I with List II :

List I

- A. Pterophyllum
B. Myxine
C. Pristis
D. Exocoetus

List II

- I. Hag fish
II. Saw fish
III. Angel fish
IV. Flying fish

Choose the correct answer from the options given below :

- (A) A-II, B-I, C-III, D-IV
(B) A-III, B-I, C-II, D-IV
(C) A-IV, B-I, C-II, D-III
(D) A-III, B-II, C-I, D-IV

185. Following are the stages of cell division :

- A. Gap 2 phase
- B. Cytokinesis
- C. Synthesis phase
- D. Karyokinesis
- E. Gap 1 phase

Choose the correct sequence of stages from the options given below :

- (A) C-E-D-A-B
- (B) E-B-D-A-C
- (C) B-D-E-A-C
- (D) E-C-A-D-B

Section B

186. Given below are two statements:

Statement I: Mitochondria and chloroplasts both double membranes bound organelles.

Statement II: Inner membrane of mitochondria is relatively less permeable, as compared chloroplast. In the light of the above statements, choose the mis appropriate answer from the options given below:

- (A) Both Statement I and Statement II are correct.
- (B) Both Statement I and Statement II are incorrect.
- (C) Statement I is correct but Statement II is incorrect.
- (D) Statement I is incorrect but Statement II is correct

187. Match List I with List II:

	List I		List II
A.	Mesozoic Era	I.	Lower invertebrates
B.	Proterozoic Era	II.	Fish & Amphibia
C.	Cenozoic Era	III.	Birds & Reptiles
D.	Paleozoic Era	IV.	Mammals

Choose the correct answer from the options given below :

- (A) A-II, B-I, C-III, D-IV
- (B) A-III, B-I, C-II, D-IV
- (C) A-I, B-II, C-IV, D-III
- (D) A-III, B-I, C-IV, D-II

188. Given below are two statements:

Statement I: Gause's competitive exclusion principle states that two closely related species competing for different resources cannot exist indefinitely.

Statement II: According to Gause's principle, during competition, the inferior will be eliminated. This may be true if resources are limiting.

In the light of the above statements, choose the correct answer from the options given below :

- (A) Both Statement I and Statement II are true.
- (B) Both Statement I and Statement II are false.
- (C) Statement I is true but Statement II is false.
- (D) Statement I is false but Statement II is true.

193. Given below are two statements:

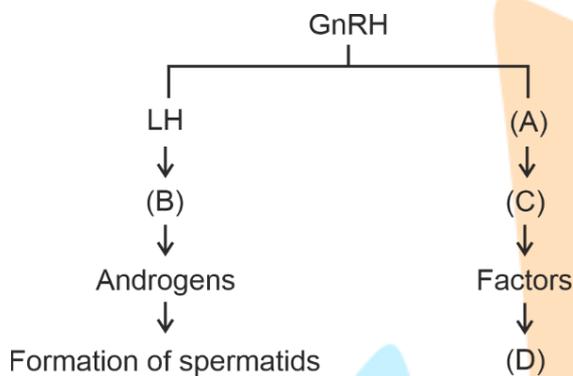
Statement I: The cerebral hemispheres are connected by nerve tract known as corpus callosum.

Statement II: The brain stem consists of the medulla oblongata, pons and cerebrum.

In the light of the above statements, choose the most appropriate answer from the options given below:

- (A) Both Statement I and Statement II are correct.
- (B) Both Statement I and Statement II are incorrect.
- (C) Statement I is correct but Statement II is incorrect.
- (D) Statement I is incorrect but Statement II is correct.

194. Identify the correct option (A), (B), (C), (D) with respect to spermatogenesis.



- (A) FSH, Leydig cells, Sertoli cells, spermiogenesis.
- (B) ICSH, Interstitial cells, Leydig cells, spermiogenesis.
- (C) FSH, Sertoli cells, Leydig cells, spermatogenesis.
- (D) ICSH, Leydig cells, Sertoli cells, spermatogenesis.

195. As per ABO blood grouping system, the blood group of father is B⁺, mother is A⁺ and child is O⁺.

Their respective genotype can be

As per ABO blood grouping system, the blood group of father is B⁺, mother is A⁺ and child is O⁺.

Their respective genotype can be

- A. IBi/IAi/ii
- B. IBIB/IAIA/ii
- C. IAIB/iIA/IBi
- D. IAi/IBi/IAi
- E. iIB/iIA/IAIB

Choose the most appropriate answer from the options given below :

- (A) A only
- (B) B only
- (C) C & B only
- (D) D & E only

196. Given below are two statements :

Statement I : Bone marrow is the main lymphoid organ where all blood cells including lymphocytes are produced.

Statement II : Both bone marrow and thymus provide micro environments for the development and maturation of T-lymphocytes.

In the light of above statements, choose the most appropriate answer from the options given below

- (A) Both Statement I and Statement II are correct.
 (B) Both Statement I and Statement II are incorrect.
 (C) Statement I is correct but Statement II is incorrect.
 (D) Statement I is incorrect but Statement II is correct.

197. Regarding catalytic cycle of an enzyme action, select the correct sequential steps :

- A. Substrate enzyme complex formation.
 B. Free enzyme ready to bind with another substrate.
 C. Release of products.
 D. Chemical bonds of the substrate broken.
 E. Substrate binding to active site.

Choose the correct answer from the options given below :

- (A) E, A, D, C, B
 (B) A, E, B, D, C
 (C) B, A, C, D, E
 (D) E, D, C, B, A

198. Match List I with List II :

	List I		List II
A.	P wave	I.	Heart muscles are electrically silent.
B.	QRS complex	II.	Depolarisation of ventricles.
C.	T wave	III.	Depolarisation of atria.
D.	T-P gap	IV.	Repolarisation of ventricles.

Choose the correct answer from the options given below :

- (A) A-I, B-III, C-IV, D-II
 (B) A-III, B-II, C-IV, D-I
 (C) A-II, B-III, C-I, D-IV
 (D) A-IV, B-II, C-I, D-III

199. Match List I with List II :

	List I		List II
A.	Exophthalmic goiter	I.	Excess secretion of cortisol, moon face & hyperglycemia.
B.	Acromegaly	II.	Hypo-secretion of thyroid hormone and stunted growth.
C.	Cushing's syndrome	III.	Hyper secretion of thyroid hormone & protruding eye balls.
D.	Cretinism	IV.	Excessive secretion of growth hormone.

Choose the correct answer from the options given below :

- (A) A-I, B-III, C-II, D-IV
 (B) A-IV, B-II, C-I, D-III
 (C) A-III, B-IV, C-II, D-I
 (D) A-III, B-IV, C-I, D-II

200. The following are the statements about non-chordates:

- A. Pharynx is perforated by gill slits.
- B. Notochord is absent.
- C. Central nervous system is dorsal.
- D. Heart is dorsal if present.
- E. Post anal tail is absent.

Choose the most appropriate answer from the options given below:

- (A) A & C only
- (B) A, B & D only
- (C) B, D & E only
- (D) B, C & D only

